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Module 3

Environmental Chemistry

Home Instructor's Guide and Assignment Booklet 3B











Science 9 Module 3: Environmental Chemistry Home Instructor's Guide and Assignment Booklet 3B Learning Technologies Branch ISBN 0-7741-2588-8

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The Learning Technologies Branch acknowledges with appreciation the Alberta Distance Learning Centre and Pembina Hills Regional Division No. 7 for their review of this Home Instructor's Guide and Assignment Booklet.

This document is intend	led for
Students	1
Teachers	1
Administrators	
Home Instructors	1
General Public	
Other	



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- · Alberta Learning, http://www.learning.gov.ab.ca
- · Learning Technologies Branch, http://www.learning.gov.ab.ca/ltb
- Learning Resources Centre, http://www.lrc.learning.gov.ab.ca

The use of the Internet is optional. Exploring the electronic information superhighway can be educational and entertaining. However, be aware that these computer networks are not censored. Students may unintentionally or purposely find articles on the Internet that may be offensive or inappropriate. As well, the sources of information are not always cited and the content may not be accurate. Therefore, students may wish to confirm facts with a second source.

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Section 2: The Problem of Pollution

In this section the student discovers the role of chemistry in assessing what substances are pollutants. The student identifies pollution sources, assesses water quality, and uses techniques of environmental monitoring. The student applies this knowledge along with risk assessment concepts in decision-making. The decision-making involves issues related to pollution and waste management.

The following materials will be needed to complete this section.

Section 2: Lesson 1

- · safety goggles
- an apron
- · latex gloves
- eight clean plastic drinking glasses of the same shape and size
- · a marking pen
- a spoon
- a ruler
- salt
- sugar

Section 2: Lesson 2

No extra materials are needed for this lesson.

Section 2: Lesson 3

No extra materials are needed for this lesson.

Section 2: Lesson 4

No extra materials are needed for this lesson.

Suggested Answers

Section 2: Lesson 4

Textbook questions 1, 4, 7, 9, and 11 of "Wrap-up: Topics 4 to 6," page 253:

- Pollution is material or energy added to the environment in large enough concentrations to cause harm to organisms or their habitat.
- 4. Macroinvertebrates are large enough to be easily collected, identified, and studied. Sensitivity to pollutants varies from species to species. They live and eat in the water, and their populations have been proven to correlate with water quality.
- 7. Heat energy is considered to be a pollutant when it is added to the natural environment in large enough quantities to cause harm to organisms or their habitat. Temperature changes can cause reproductive problems and alterations to oxygen and chemical concentrations in the water. Each species has a range of tolerance for temperature. If this range is exceeded, the species will do poorly or die out.

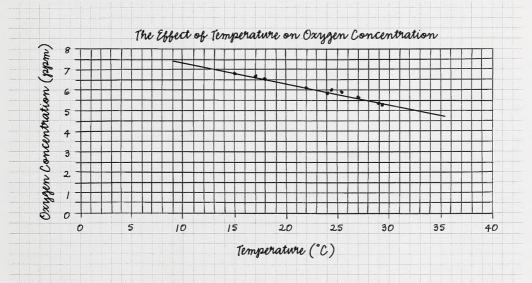
- Recycling reduces the overall demand for energy, water, and materials. It reduces the amount of pollution generated in the production of goods, and it decreases the amount of material that must be stored in a landfill.
- Mercury is used in fluorescent light bulbs as "ballast." Since mercury is a heavy metal with toxic characteristics, the light bulbs that contain mercury become hazardous waste when they are discarded.

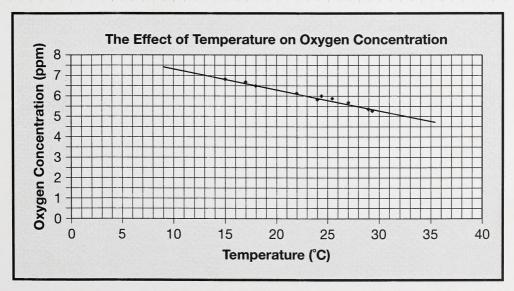
Module Review

- 1. Textbook questions 1, 4, 5, 8 to 11, 15, 19, 20, 25, and 33 from "Understanding Key Concepts," pages 258 to 260:
 - 1. Each of these elements is required for bone formation.
 - 4. Plants bioaccumulate minerals from their substrate. These essential nutrients are then passed on to animals.
 - 5. Bacteria in the soil decompose dead matter or waste matter to make the nutrients available for plant growth. Bacteria also convert nitrogen into compounds that plants can use (nitrogen fixation).
 - 8. Fungicides are used to control fungi, herbicides control weeds, and insecticides control insects.
 - 9. DDT was a cheap and effective chemical for killing disease-bearing and crop-damaging insects. It was considered to be a safe chemical when it was introduced.
 - 10. DDT is a stable, non-excretable toxin that bioaccumulates in organisms and biomagnifies through the food web. Its breakdown product, DDE, is now considered to be a toxin as well. DDT is still in use in several developing countries.
 - 11. A poison causes illness or death when it's absorbed or ingested. Toxins are protein molecules, produced by organisms, that cause illness or death. Not all poisons are toxins. *Toxic* refers to the result (illness or death), not to the chemical causing the result. The term *toxic* can therefore be applied to the results produced by both poisons and toxins.
 - 15. Sulfur oxides released by industry and transport react with water in the atmosphere to produce sulfuric acid (H_2SO_4) and sulfurous acid (H_2SO_3) . Nitrogen oxides produce nitric acid (HNO_3) and nitrous acid (HNO_2) . Carbon dioxide produces carbonic acid (H_2CO_3) .
 - 19. Eastern Canada is most affected by acid precipitation. It is the most industrialized area, and it is located downwind of major manufacturing areas in the United States. Eastern Canada is located on the inert rocks and soils of the Canadian Shield, which is not naturally buffered by limestone.
 - 20. Catalytic converters are used on cars to decrease the amount of nitrogen oxides and carbon monoxide released by burning fossil fuel. The converters facilitate more complete fuel combustion—this decreases the release of acid precipitation precursors and carbon particulates.
 - 25. Non-persistent pollutants are relatively easy to biodegrade into non-toxic breakdown products.
 - **33.** Global wind and water currents carry pollutants into polar regions. Cold temperatures slow or prevent their breakdown, which allows more time for the pollutants to biomagnify through the food web.

- 2. a. Here is a checklist that can be used to assess student graphs:
 - The graph has a specific title.
 - The graph is in pencil or is computer generated.
 - Lines (such as axes) are straight.
 - The *x*-axis is labelled and the units are printed.
 - The y-axis is labelled and has printed units.
 - The *x* and *y* axis scales are appropriate.
 - Data is placed appropriately on the grid.
 - The line of best fit is included properly.

Student graphs should be similar to one of the following graphs. The second graph is made by a computer spreadsheet.





- **b.** As the water temperature increases, the oxygen concentration in the water decreases.
- c. Answers will vary somewhat. The value is read from the line of best fit. Based on interpolation, the oxygen concentration at 20° C would be 6.3 ppm.
- **d.** Answers will vary somewhat. The value is again read from the line of best fit, which may have to be extended. Based on extrapolation, the oxygen concentration at 9° C would be 7.5 ppm.
- **e.** Yes, there is a very strong relationship between water temperature and oxygen concentration. All the data points are near to the line of best fit.

Note: Dissolved oxygen does not refer to the oxygen atom in each water molecule. The oxygen atoms within water molecules are not available for "breathing." The dissolved oxygen in water is actually composed of a very small number of molecules. How small is this number? There are at most just 10 molecules of oxygen interspersed among every million molecules of water. This dissolved form of oxygen is what is available to aquatic organisms.

ASSIGNMENT BOOKLET 3B

Science 9

Module 3: Section 2 Assignment and Final Module Assignment

Home Instructor's and Student's	Comments:		
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		verify	
		Please	Grading:
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	Name		
Teacher's Comments			
			Teacher's Signature

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- Are all the assignments completed? If not, explain why.
- Has your work been reread to ensure accuracy in spelling and details?
- Is the booklet cover filled out and the correct module label attached?

MAILING

- 1. Do not enclose letters with your Assignment Booklets. Send all letters in a separate envelope.
- 2. Put your Assignment Booklet in an envelope and take it to the post office and have it weighed. Attach sufficient postage and seal the envelope.

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- 2. All faxing costs are the responsibility of the sender.

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Module 3

Environmental Chemistry *Assignment Booklet 3B*





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FOR TEACHER'S USE ONLY

Summary

	Total Possible Marks	Your Mark
Section 2 Assignment	37	
Final Module Assignment	70	
	107	

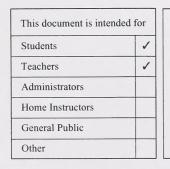
Teacher's Comments

Science 9
Module 3: Environmental Chemistry
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Section 2 Assignment and Final Module Assignment
Learning Technologies Branch

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ASSIGNMENT BOOKLET 3B SCIENCE 9: MODULE 3 SECTION 2 ASSIGNMENT AND FINAL MODULE ASSIGNMENT

Your mark for this module will be determined by how well you do your assignments.

This Assignment Booklet is worth 107 marks out of the total 173 marks for the assignments in Module 3. The value of each assignment and each question is stated in the left margin.

	Wo	ork slowly and carefully. If you have difficulty, go back and review the appropriate topic.			
	Ве	sure to proofread your answers carefully.			
37	Se	ection 2 Assignment: The Problem of Pollution			
31)		Read all parts of your assignment carefully and record your answers in the appropriate places.			
3	1.	Some students debated the issue of putting chlorine in water. In your opinion, is chlorinating water worth the cancer risk? Support your opinion.			
2	2.	Is a dioxin concentration of 0.04 ppm below the LD50 standard? Explain.			
3	3.	Some people feel that "dilution is the solution to pollution." Could toxins found in air, water, or soil in concentrations well below LD50 ever become very harmful to you or other organisms? Explain.			



Answer questions 4 to 6 by using the information and data supplied for "Inquiry Investigation 3I: Testing Water Quality" in Section 2: Lesson 2 of the Student Module Booklet.

Refer to the corresponding investigations on pages 225 to 229 of the textbook as well

3	4.	Was the tested fertilizer meant to grow leaves, roots, or flowers? Explain.
1	5.	Refer to all the pond water data. a. Would you consider the water to be polluted?
		b. Provide three specific measured data values to support your opinion. Include units. For each data value, interpret whether it is high or low. Also indicate any effect on aquatic life.
1		data value:
2		interpretation/effect:
1		• data value:
2		interpretation/effect:

1	• data value:
2	interpretation/effect:
	6. The amount of dissolved oxygen in a body of water would be affected by any of the following:
	 an increase in the water temperature an increase in the surface area of the water in contact with air an increase in the number of plants in the water an increase in the number of animals in the water
2	a. Which of these changes would lead to a decrease in dissolved oxygen?
2	b. Which of these changes would lead to an increase in dissolved oxygen?
	Return to page 47 of the Student Module Booklet and continue with Lesson 2.
	7. Refer to both the data table and the drawing on page 235 of the textbook. You will see the number of organisms detected along a stream. Use this information to answer the following questions.
1	a. Classify the water at site C as high quality, medium quality, or low quality. Explain.

1	ł	Aquatic worms are tolerant of low oxygen conditions. Site E has a fairly high population of aquatic worms. Do you consider site E to be polluted? Explain.
2	c	c. Consider the populations of dragonfly nymphs at sites C and D. A set of rapids is located between these two sites. Explain the difference in the dragonfly populations at these two sites.
1	d	I. It has been determined that both the sewage outlet and the agricultural areas along the river basin contribute to the river's water quality. What is the major difference between these two sources?
	e e e e e e e e e e e e e e e e e e e	Return to page 51 of the Student Module Booklet and continue with Lesson 3.
2	8.	Could the people living a traditional lifestyle in the high Arctic be classified as bioindicators of industrial pollution? Explain.
1	9.	Name a pollutant that attacks the ozone layer.
1	10.	Why is the ozone layer important to living things?

2	11. A grade 9 student was describing her science class to her younger brother. She spoke about pollutants spreading in the environment. Her brother suggested a simple way to get rid of waste liquids, such as used motor oil and paint thinners. He suggested that such liquids should simply be thrown out in a pit and allowed to seep into the ground. Give two reasons why this would not be a good idea.
1)	12. Circle the letter of the correct response.
	A. salts B. solvents C. sugars D. solids Return to page 61 of the Student Module Booklet and continue with the Module Summary.
(70)	Final Module Assignment
(10)	Read all parts of your assignment carefully and record your answers in the appropriate places.
1	1. The use of chlorofluorocarbons (CFCs) has been banned around the globe for many years. Explain why the ozone layer is still thinning.
3	2. "The dose, the time period, and the individual determine the toxicity." Explain this phrase.

1	3.	It is usually a bigger problem when a point source discharges into ground water than it is when a point source discharges into surface water. Explain why.
1)	4.	Warnings on the labels of many types of glue advise you to avoid skin contact with the glue, and they also advise you to use the glue in a well-ventilated area. Explain why these warnings are necessary.
	5.	Excessive runoff from fields and feedlots affects the populations of the caddis fly larvae and gilled snails in a nearby stream.
1		a. What would happen to the populations of these two species as a result of this excessive runoff?
1		b. What environmental factor would cause the population changes? Be specific.
1		c. Name the chain of events that would lead to the change in this environmental factor.
2	6.	Coastal towns and cities often release untreated household sewage and industrial sewage directly into the ocean. Briefly describe two natural mechanisms that would reduce the resulting pollution.

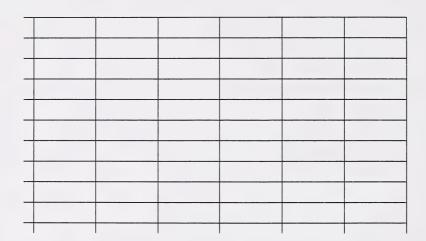
2	7. Water quality tests can play an important part in assessing the level of pollutants in the environment. Provide two reasons why testing water quality can be particularly useful to determine environmental health.
	8. Everyone is responsible for some pollution.
2	a. Name two different, common activities that contribute to surface water pollution.
1	b. What could you do to prevent or reduce one of these pollution sources?
2	9. Determine how much solanine a 60 kg person would have to consume to risk a 50% chance of death. Remember: 1 ppm is equivalent to 1 mg/kg. Show your work including units



10. Refer to the data table on page 260 of the textbook. Read the information in the paragraph above the table. The information shows the relation between the number of resistant species and the number of insecticides in use. Data was recorded in the given area for over 40 years.

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a. Graph both sets of data on the same axes. Place the year on the horizontal axis and the number on the vertical axis. Be sure to meet all the appropriate graphing criteria.



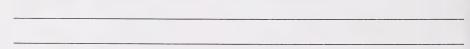
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(3)

As the years of insecticide use increase, the number of resistant species
and the number of insecticides in use
The number of resistant species changes at a

-	$\overline{}$	•
-	-4	7
1	7	

c. What is likely the reason for the difference in the rate of change?



_____ rate than the number of insecticides in use.

(3)		fight these pests? Back your opinion with two specific reasons. Opinion:
		One reason:
		Your other reason:
2	11.	The recommended daily allowance for zinc is 12 to 25 mg per day. Consuming 150 mg or more zinc per day can cause a copper deficiency. Some manufacturers recommend using zinc lozenges to soothe a sore throat, at a rate of about 13 mg of zinc every two hours. Explain why using this recommended dosage of zinc throat lozenges could lead to anemia, which is a low red blood cell count.
<u></u>	12.	Your lawn is looking less healthy than it has for several years. Your neighbour offers to
ತ		give you his bag of 7-7-15 fertilizer. Should you accept his offer based on what fertilizer would help your lawn? Explain.

2	13.	There has been a large spill of sulfuric acid into a lake. A week later, tests show that the pH of the lake is nearly normal. What could make these results possible? Reason 1:
		Reason 2:
4)	14.	Polychlorinated biphenyls (PCBs) are stable, non-excretable, fat-soluble toxins. Your local government has proposed building a high-temperature incineration plant for PCBs and similar chemicals. The plant would process chemicals from western Canada and the United States. Provide two reasons in favour of building this plant and two reasons against. The reasons must be specific, and each one must be different. Pro 1:
		Pro 2: Con 1:

		Con 2:		
1)	15.	"Any chemical is safe is incorrect.	f it is consumed in small	enough doses." Explain why this statemen
5	16.	Explain what LD50 is a resistance.	nd describe how it relate	es to the development of pesticide
	17.	Use these scientific terr • aquifer • permeable • surface water • transpire • clay	 dispersal leachate ground water sand bioaccumulate 	 impermeable nitrates and E. coli pesticides and plant nutrients sulfur and nitrogen oxides are diluted
				f pollution in the environment. Fill in each entences meaningful. Note that not all the
1		a. The rate of affected by soil por	osity, soil type, rock type	of a chemical in ground water is e, and slope.

2		b.	A(An)	results from ground water being held
			above a layer of	material.
2		c.		_ is the polluted liquid that flows downward
			through a landfill. In order to keep this	liquid out of ground water, the landfill must be
			lined with	
1		d.	Untreated household sewage contains _ pollutants.	as
3		e.	The run-off from agricultural land likel	ly contains
				Some of these pollutants combine with
				_ to cause algal blooms. Others enter the food
			chain and	to harmful levels.
1		f.		_ enter the water cycle to form acid rain.
(3)	18.	Cla	assify the following solutions as acid, ba	se, or neutral.
		a.	pH 9.3	b. pH 1.5
		c.	pH 2.5	d. pH 6.9
		e.	pH 7.0	f. pH 12.5
3	19.	Lis	at the solutions in question 18 from the n	nost basic to the most acidic.
		_		

ASSIGNMENT BOOKLET DECLARATIONS

The school you are registered with may require you to submit this signed form with your Assignment Booklet.

The Student's Declaration is to be signed by the student. If the student is under 16, the Supervisor's Declaration may need to be signed by the supervisor, who is usually a home instructor, teacher, or home-schooling coordinator. Failure to complete this page may invalidate the assignment results. Please contact your school and ask if this completed form is required.

STUDENT'S DECLARATION

SUPERVISOR'S DECLARATION

- I have followed the instructions outlined in the Student Module Booklet.
- I have completed the activities to prepare myself for the assignments in this Assignment Booklet.
- I completed the assignments in this Assignment Booklet by myself.

Student's Signature

	g activities completed byStudent's Name
lso certify that to the best of my knowledge the assignments in this Assignment Booklet were complete dependently by this student.	
	Supervisor's Signature
	ents or observations regarding this module, write them in
ne following space.	